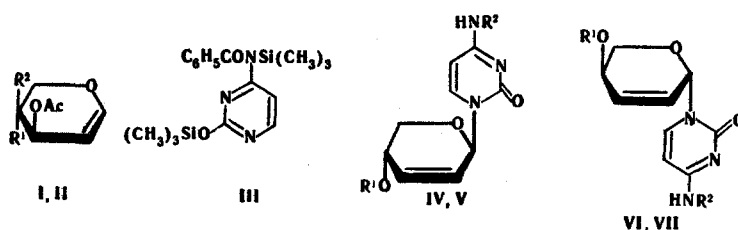


We found that the reaction of 3,4-di-O-D-acetyl-xylal (I) and 3,4-di-O-acetyl-L-arabinal (II) with N⁴,O-bis(trimethylsilyl)-N⁴-benzoylcytosine (III) in dichloroethane in the presence of SnCl₄ leads to the formation of 1-(4-O-acetyl-2,3-didesoxy-β-D-glyceropent-2-enopyranosyl)-N⁴-benzoylcytosine (IV) in a yield of ~40%, mp 225-226° (from methanol), $[\alpha]_D^{20} +104^\circ$ (s, 2, MeOH) and 1-(4-O-acetyl-2,3-didesoxy-β-L-glyceropent-2-enopyranosyl)-N⁴-benzoylcytosine (VI), mp 225-226°C (from methanol), $[\alpha]_D^{20} -107^\circ$ (s, 2, MeOH), respectively. Nucleosides IV and VI have identical IR, UV, PMR and mass spectra, which together with the identical in value but opposite in sign Cotton effect on the rotational dispersion curves indicates that they form an enantiomeric pair.



I R¹=OAc, R²=H; II R¹=H, R²=OAc; IV R¹=Ac, R²=C₆H₅CO; V R¹=R²=H; VI R¹=Ac, R²=C₆H₅CO; VII R¹=R²=H.

The treatment of compound IV and VI with a methanolic solution of ammonia gave 1-(2,3-dideoxy-β-D-glyceropent-2-enopyranosyl) cytosine (V), mp 214-215°C (from water), $[\alpha]_D^{20} +75^\circ$ (s 2, H₂O), and 1-2,3-dideoxy-β-L-glyceropent-2-enopyranosyl)cytosine (VII), mp 214-215°C (from water), $[\alpha]_D^{20} -68^\circ$ (s 2, H₂O), respectively. The structure of the compounds obtained was confirmed by the data of IR, UV, PMR, rotational dispersion and mass spectra, and by elementary analysis.